

Evaluation of Inductively Coupled Plasma Tandem Mass Spectrometry for Determination of As in Agricultural Inputs with High REE Contents

Raquel C. Machado,^{*,a,b} Alex Virgílio,^a Clarice D. B. Amaral,^{a,b} Daniela Schiavo,^c
Joaquim A. Nóbrega^a and Ana Rita A. Nogueira^b

^aGrupo de Análise Instrumental Aplicada, Departamento de Química, Universidade Federal de São Carlos, 13565-905 São Carlos-SP, Brazil

^bEmbrapa Pecuária Sudeste, 13560-970 São Carlos-SP, Brazil

^cAgilent Technologies, 06455-000 Barueri-SP, Brazil

In the present work, an inductively coupled plasma-tandem mass spectrometer (ICP-MS/MS) operating in single quadrupole and mass shift modes were evaluated for As determination in mineral fertilizers and agricultural gypsum with high contents of rare earth elements. For MS/MS mode, oxygen was the gas introduced into the octopole reaction system (ORS³). Samples were prepared by microwave-assisted digestion with diluted nitric acid solution. The accuracy was checked by analysis of a fertilizer certified reference material (NIST SRM 695) and by addition and recovery experiments. When operating in single quadrupole mode, recoveries ranged from 59 to 151%; while values obtained by MS/MS mode varied from 81 to 105% when 0.30 mL min⁻¹ O₂ was introduced into the ORS³. Limits of detection for As⁺ in single quadrupole and AsO⁺ in MS/MS mass shift mode were 6 and 9 ng L⁻¹, respectively.

Keywords: arsenic, double charged interference, fertilizer, gypsum, ICP-MS/MS

Introduction

The growth of food production over the years contributed to increasing the application of agricultural inputs to soils. In addition to nutrients, agricultural inputs may also be a source of toxic elements as As, Cd, Cr, Hg and Pb at trace levels, which can accumulate in the soil and enter into the food chain, therefore their concentrations need to be monitored.¹ The toxic effects of inorganic arsenic exposure on human health are well reported in the literature and some of the negative secondary effects on lung, liver, skin and bladder cancers, diabetes, cardiovascular and neurological diseases are among them.²

The minerals fertilizers (i.e., phosphate fertilizers) and soil conditioners as agricultural gypsum, a co-product of phosphate fertilizers processing plants, are inputs that contain macronutrients (Ca, Mg, N, P, and S), micronutrients (such as Fe and Si), rare earth elements (REEs, such as Ce, Dy, Er, Eu, Gd, Ho, La, Lu, Nd, Pr, Sc, Sm, Tb, Tm, Y, and Yb) and, in some cases, toxic elements (As, Cd, Hg and Pb) in their composition.

Spectral interferences can impair the determination of As in this type of samples.³⁻⁵

Inductively coupled plasma mass spectrometry (ICP-MS) has been employed for trace elements determination, such as As, due to its high sensitivity, multielemental capability indicating the possibility of carrying out the determination of several elements in the same run, wide linear range and lower limits of detection.^{6,7} Accurate ⁷⁵As⁺ determination is usually a challenge because of the well-known isobaric overlaps caused by polyatomic species, such as ⁴⁰Ar³⁵Cl⁺, originated from the presence of elements of the matrix in the argon plasma.⁸⁻¹⁰

In addition to polyatomic interferences, double charged species may also affect the accurate determination of As. Some of these species, such as ¹⁵⁰Nd²⁺ and ¹⁵⁰Sm²⁺, are easily produced in plasma when the sample contains REEs which can overlap the As⁺ isotope at *m/z* 75.^{8,9} Correction equations,¹¹ collision and reaction interface (CRI),¹² and dynamic reaction cell (DRC)¹³ are useful strategies to overcome these interferences. The recently proposed ICP-MS/MS, designed with two quadrupoles in-between an octopole-based collision/reaction cell (ORS³), presents an attractive configuration that provides more efficient

*e-mail: raquelcm.quim@gmail.com

reaction/collision processes and better conditions to deal with spectral overlaps.¹⁴⁻¹⁹

Amaral *et al.*²⁰ verified the effectiveness of ICP-MS/MS with O₂ into the ORS³ for the elimination of interferences caused by chloride ions ⁴⁰Ar³⁵Cl⁺, ³⁵Cl¹⁶OH⁺ and ³⁵Cl¹⁶O⁺ on As⁺, Cr⁺ and V⁺ determination, respectively. The analytes were determined by their oxides with good accuracy, precision and sensitivity. Flórez *et al.*²¹ exploited a mixture of CH₃F/He (1/9) as reaction gas in single quadrupole ICP-MS to determine As from monitoring at ratio *m/z* 89 (by reaction product AsCH₂⁺ between As⁺ and CH₃F) in biological fluids that have high Cl concentration. Bolea-Fernandez *et al.*²² determined As and Se interference free by ICP-MS/MS using CH₃F (a mixture of 10% CH₃F and 90% He) as reaction gas, via AsCH₂⁺ and SeCH₂⁺ reaction products in different plant, animal and environmental reference materials.

Jackson *et al.*²³ used an ICP-MS/MS configuration with collision (He) and reaction (H₂) gases pressurized into the ORS³ in single quadrupole mode and O₂ in MS/MS mass shift mode to determine As and Se in two certified reference materials (CRM) of plant tissues, both with high REEs contents. The authors noticed that the formation of AsO⁺ and SeO⁺ and the rejection of non-target ions in MS/MS mode were required in order to gain accuracy. In another study, Jackson²⁴ employed an ion chromatography-ICP-MS/MS mass shift mode to determine arsenocholine (AsC), arsenobetaine (AsB), dimethylarsine (DMA), monomethylarsine (MMA), As^{III} and As^V in CRMs of urine samples and the method developed provided better detection limits, high sensitivity and sample throughput when compared to others methods.

Inorganic arsenic was determined by hydride generation (HG-ICP-MS) and high performance liquid chromatography (HPLC-ICP-MS) in rice product samples using single quadrupole and MS/MS mass shift mode, with O₂ pressurized into the ORS³. Pétursdóttir *et al.*²⁵ compared the results obtained by both methods and suggested that HPLC-ICP-MS can be replaced by simple HG system which provide As concentrations in the sample extract from low µg L⁻¹ levels.

Many countries regulate the maximum limit for As concentration in agricultural inputs. In Canada, the Federal Fertilizers Act regulates the maximum concentration as 75 mg kg⁻¹. In the United States, this concentration changes in different states, ranging from 13 to 112 mg kg⁻¹ depending on the composition of inputs. In Brazil, the Ministry of Agriculture, Livestock and Food Supply establishes the maximum allowed limits for As in fertilizers as 4,000 mg kg⁻¹.²⁶⁻²⁸

In the present work, the performance of an ICP-MS/MS operating in both single quadrupole and MS/MS mass shift

mode, using oxygen as a reaction gas, was evaluated to overcome REEs based interferences on As determination in mineral fertilizers and agricultural gypsum samples.

Experimental

Reagents, solutions and samples

All materials employed were decontaminated by immersion in 10% (v/v) HNO₃ for at least 24 h and rinsed with distilled-deionized water (resistivity > 18.2 MΩ cm), obtained from a Milli-Q[®] Water System (Millipore, Billerica, MA, USA). Standard reference solutions and spiked samples were prepared from a 1000 mg L⁻¹ of As (Qhemis, São Paulo, SP, Brazil) stock solution. For sample digestion, HNO₃ obtained from a sub-boiling distillation system (Milestone, Sorisole, Italy) and H₂O₂ 30% (m/m) (Qhemis) were used.

Samples of mineral fertilizers and agricultural gypsum from different regions were supplied by the National Laboratory for Agriculture (Goiânia, GO, Brazil) and Embrapa Southeast Livestock (São Carlos, SP, Brazil). A fertilizer certified reference material, SRM 695, from the National Institute of Standards and Technology (NIST, Gaithersburg, MD, USA) was used to check the accuracy.

Instrumentation

All measurements were carried out in an Agilent 8800 inductively coupled plasma tandem mass spectrometer ICP-MS/MS (Agilent Technologies, Tokyo, Japan) equipped with ORS³ located between two quadrupole mass analyzers. Liquid argon was provided by White Martins (Sertãozinho, SP, Brazil). Pure oxygen (≥ 99.999%, Air Products, São Paulo, SP, Brazil) was used as cell gas. The single quadrupole and MS/MS mass shift modes were evaluated. The operational parameters are presented in Table 1. Samples were digested using an Ethos 1 microwave oven (Milestone) and were analyzed in triplicate.

Sample preparation

Mineral fertilizers and agricultural gypsum sample masses (ca. 200 mg) were accurately weighed and aliquots of 6 mL of HNO₃ 7 mol L⁻¹ and 2 mL of H₂O₂ 30% (m/m) were added to the vessels. The microwave oven two-step heating program was applied as follows: (i) 20 min ramp to 200 °C; (ii) 20 min plateau at 200 °C. After cooling down, the digested samples were transferred to polypropylene flasks and volumes were made up to 50.0 mL with distilled-deionized water. All samples were digested in triplicate.

Table 1. Operating conditions for the Agilent 8800 ICP-MS/MS

Parameter	Operating condition		
RF applied power / W	1550		
Plasma gas flow rate / (L min ⁻¹)	18		
Auxiliary gas flow rate / (L min ⁻¹)	1.8		
Carrier gas flow rate / (L min ⁻¹)	1.08		
Nebulizer	concentric nebulizer		
Spray chamber	Scott type-double pass		
Sampling depth / mm	8		
Number of replicates	3		
Mass/charged ratio selected-Q ₁	75 (As ⁺), 150 (Nd ⁺ , Sm ⁺)		
Mass/charged ratio selected-Q ₂	91 (AsO ⁺), 166 (NdO ⁺ , SmO ⁺)		
Operating mode	single quadrupole	MS/MS	
Cell gas	No gas	O ₂	
O ₂ flow rate / (mL min ⁻¹)	0	0.30	0.50

MS/MS: tandem mass spectrometry.

Results and Discussion

Double charged ion monitoring and determination of elements at m/z 150 in agricultural inputs

The As⁺ isotope at m/z 75 is overlapped by double charged species formed by ¹⁵⁰Sm²⁺ or ¹⁵⁰Nd²⁺ present in mineral fertilizer, agricultural gypsum and SRM 695 samples, which can cause an overestimation of the ⁷⁵As⁺ concentrations. To check the presence of interfering species, the net signal intensities at m/z 150 were monitored in the blank solution 1% (v/v) HNO₃ and in the samples using single quadrupole mode (Figure 1). As it can be seen, the signal intensities of agricultural gypsum samples are 7 to 400-fold higher than those observed for mineral fertilizers and SRM 695, respectively.

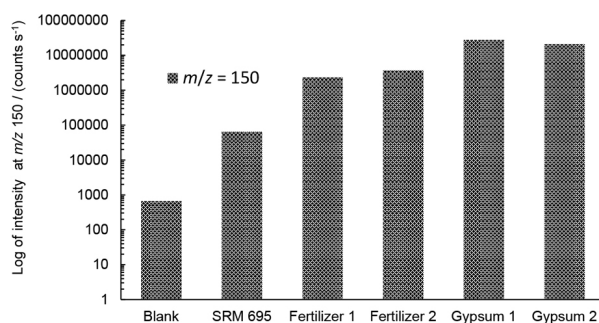


Figure 1. Net signal intensities at m/z 150 for blank solution 1% (v/v) HNO₃, SRM (695), fertilizer and gypsum samples obtained in single quadrupole mode.

In order to check if the high intensities observed at m/z 150 were originated from REE ions or oxide ions

as ¹³²Ba¹⁸O⁺ and ¹³⁴Ba¹⁶O⁺, since Ba is present at high concentrations, the concentrations of ¹⁵⁰Nd⁺ and ¹⁵⁰Sm⁺ in the samples were determined in the single quadrupole and MS/MS mass shift modes as ¹⁵⁰Nd¹⁶O⁺ and ¹⁵⁰Sm¹⁶O⁺ by introducing 0.30 mL min⁻¹ O₂ into the ORS³. Results are shown in Table 2. It can be observed that in all samples, the determined concentrations for elements at m/z 150 (¹⁵⁰Nd⁺ or ¹⁵⁰Sm⁺) were higher than 100 mg kg⁻¹ and, for agricultural gypsum samples, concentrations 8 to 11-fold higher than in mineral fertilizers were obtained. No statistical differences were found at a 95% confidence level (*t*-test) when both operation modes were considered. Addition and recovery experiments were performed for evaluating the accuracy of the measurements at m/z 150 by adding 40 µg L⁻¹ of Nd and Sm to fertilizer and gypsum digests. Recoveries were between 82 to 89% and 90 to 93% when operating in single quadrupole and MS/MS mass shift modes, respectively.

Table 2. Determination of ¹⁵⁰Nd⁺ and ¹⁵⁰Sm⁺ (mean ± standard deviation, n = 3) in mineral fertilizers and agricultural gypsum in single quadrupole and MS/MS mass shift modes (0.30 mL min⁻¹ O₂ into the ORS³)

Sample	Single quadrupole mode (¹⁵⁰ Nd ⁺ , ¹⁵⁰ Sm ⁺)	MS/MS mass shift mode (¹⁵⁰ Nd ¹⁶ O ⁺ , ¹⁵⁰ Sm ¹⁶ O ⁺)
	Determined concentration / (mg kg ⁻¹)	Determined concentration / (mg kg ⁻¹)
Fertilizer 1	213 ± 4	231 ± 4
Fertilizer 2	126 ± 8	126 ± 5
Gypsum 1	1459 ± 57	1433 ± 47
Gypsum 2	1037 ± 35	1004 ± 37

MS/MS: tandem mass spectrometry.

Effect of double charged ions ¹⁵⁰Sm²⁺ and ¹⁵⁰Nd²⁺ on ⁷⁵As⁺ analytical signal intensities

The effects of ¹⁵⁰Sm²⁺ and ¹⁵⁰Nd²⁺ on ⁷⁵As⁺ net signal intensities were evaluated in solutions containing 1.0 µg L⁻¹ As in the presence of 0, 5, 10, 20, 40, 70 and 100 µg L⁻¹ Nd and Sm (Figure 2). Signal intensities were obtained in single quadrupole (⁷⁵As⁺) and MS/MS mass shift modes (⁷⁵As¹⁶O⁺, 0.30 mL min⁻¹ into the ORS³) and normalized to 1 when Nd and Sm were absent. It is important to emphasize that in the case of single quadrupole measurements, ⁷⁵As⁺ was the monitored specie while in MS/MS mass shift mode, oxides were easily formed into the ORS³ when O₂ was introduced, thus ⁷⁵As¹⁶O⁺ was selected for measurements at the second quadrupole.

As can be seen in Figure 2, when Nd and Sm concentrations increased, the signal at m/z 75 in single quadrupole also increased up to 60% due to spectral interferences caused by double charged species. When the

MS/MS mass shift mode was applied, the observed gain was lower than 7%, thus evidencing that the addition of oxygen into the ORS³ cell, the mass shift from ⁷⁵As⁺ to ⁷⁵As¹⁶O⁺ and the selection of the oxide specie in the second quadrupole (Q₂) are efficient strategies to eliminate double charged ions interferences caused by REEs. It is also important to consider that possible interfering species, such as ⁹¹Zr⁺, at the same mass/charge ratio of ⁷⁵As¹⁶O⁺ are rejected in the first quadrupole (Q₁) by allowing only *m/z* 75 ions to pass through.

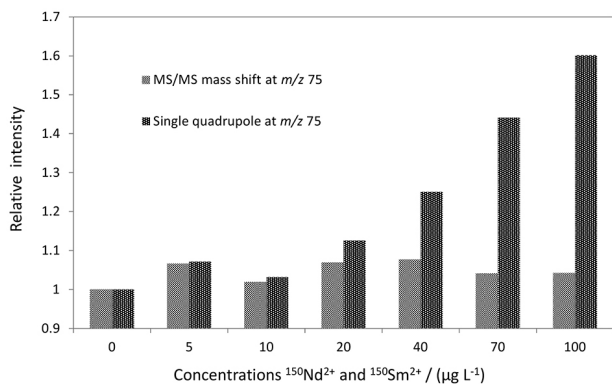


Figure 2. Relative signal intensities on *m/z* 75 for 1 µg L⁻¹ As and increasing concentrations of Sm and Nd in single quadrupole and MS/MS mass shift modes.

Analytical performance and accuracy

The analytical performance of ICP-MS/MS for As determination was evaluated for ⁷⁵As⁺ (single quadrupole mode) and ⁷⁵As¹⁶O⁺ (MS/MS mass shift mode). Limits of detection and sensitivities for As in single quadrupole and MS/MS mass shift modes are shown in Table 3. When oxygen gas was introduced into the ORS³, it was observed a sensitivity reduction and higher limit of detection (LOD) values when compared to single quadrupole mode in the absence of cell gas, probably due to the ion beam defocusing and the efficiency of the reaction processes. Despite the lower background signals obtained in MS/MS mode, the effect on sensitivity is even more pronounced when the oxygen flow rate was increased from 0.30 to 0.50 mL min⁻¹.

The ICP-MS operated in single quadrupole mode and MS/MS mass shift mode (with different oxygen

flow rates into the ORS³) was evaluated for correction of interferences at *m/z* 75. The accuracy for As determination in mineral fertilizer samples was evaluated using NIST SRM 695 (Table 4). Using the unpaired *t*-test, no significant differences between the determined and certified values at a 95% confidence level were observed and the recoveries found varied from 93 to 95% in all cases. It can be inferred that no significant improvement in accuracy was achieved using MS/MS mass shift mode with O₂ into the ORS³, which can be explained considering the low net signal intensity for *m/z* 150 (Figure 1) in this particular material.

In order to evaluate the accuracy in samples containing higher contents of elements that produce interfering double charged species, addition and recovery experiments for As at two levels (Table 4) were performed for agricultural gypsum and fertilizer samples. Recoveries obtained in single quadrupole mode ranged from 59 to 151%, while values obtained by MS/MS mass shift mode varied from 81 to 105% and 81 to 107% for O₂ flow rates of 0.30 and 0.50 mL min⁻¹, respectively. Recoveries observed in single quadrupole mode presented deviations higher than 20%, which are related to overlaps caused by double charged ions on ⁷⁵As⁺.

The As concentrations in minerals fertilizers and agricultural gypsum were determined in different instrumental conditions (Table 5). In single quadrupole mode determined concentrations are higher than those determined by MS/MS mass shift mode and this difference is more pronounced in agricultural gypsum samples, which present concentrations of elements at *m/z* 150 (¹⁵⁰Nd⁺ and ¹⁵⁰Sm⁺) up to 10-fold higher than those found in fertilizers. A paired *t*-test was used to check the agreement of results obtained by two oxygen gas flow rates (0.30 and 0.50 mL min⁻¹) and no statistical differences were found at a 95% confidence level. It was possible to verify the effectiveness of MS/MS mass shift mode using O₂ into the ORS³ to overcome double charged ions interferences on ⁷⁵As⁺ determination.

Conclusions

The use of MS/MS mass shift mode was effective to remove double charged interferences from REEs on As determination in mineral fertilizers and agricultural gypsum

Table 3. Limits of detection (LOD) and sensitivities for ⁷⁵As⁺ in single quadrupole and MS/MS modes with different oxygen gas flow rates

Single quadrupole mode (⁷⁵ As ⁺)		0.30 mL min ⁻¹ O ₂ (⁷⁵ As ¹⁶ O ⁺)		0.50 mL min ⁻¹ O ₂ (⁷⁵ As ¹⁶ O ⁺)	
LOD / (ng L ⁻¹)	Sensitivity ^a	LOD / (ng L ⁻¹)	Sensitivity ^a	LOD / (ng L ⁻¹)	Sensitivity ^a
6.0	22869	9.0	9452	40.0	6273

^aDefined as the slope of analytical calibration curves (counts L s⁻¹ µg⁻¹).

Table 4. Determination of As (mean \pm standard deviation, n = 3) in fertilizer (SRM 695) and in spiked samples of agricultural gypsum and mineral fertilizer in single quadrupole mode and MS/MS mass shift mode with different oxygen gas flow rates

Sample	Single quad mode ($^{75}\text{As}^+$)			MS/MS 0.30 mL min $^{-1}$ O_2 ($^{75}\text{As}^{16}\text{O}^+$)		MS/MS 0.50 mL min $^{-1}$ O_2 ($^{75}\text{As}^{16}\text{O}^+$)	
	Certified value / ($\mu\text{g L}^{-1}$)	Determined / (mg kg^{-1})	Recovery / %	Determined / (mg kg^{-1})	Recovery / %	Determined / (mg kg^{-1})	Recovery / %
SRM 695	200.0 \pm 5	187.4 \pm 12.5	94	190.7 \pm 17.9	95	186.5 \pm 9.79	93
	Added / ($\mu\text{g L}^{-1}$)	Determined / ($\mu\text{g L}^{-1}$)	Recovery / %	Determined / ($\mu\text{g L}^{-1}$)	Recovery / %	Determined / ($\mu\text{g L}^{-1}$)	Recovery / %
Fertilizer 1	5.0	5.8 \pm 0.3	116	4.0 \pm 0.2	81	4.1 \pm 0.3	81
	10.0	12.8 \pm 2.1	128	9.5 \pm 1.3	95	9.5 \pm 1.3	95
Gypsum 1	0.50	0.76 \pm 0.13	151	0.53 \pm 0.02	105	0.54 \pm 0.02	107
	1.0	0.59 \pm 0.51	59	0.88 \pm 0.09	88	0.91 \pm 0.10	91

Table 5. Determination in mg kg^{-1} of As (mean \pm standard deviation, n = 3) in mineral fertilizers and agricultural gypsum using single quadrupole and MS/MS mass shift modes with different oxygen gas flow rates

Sample	Single quadrupole	MS/MS 0.30 mL min $^{-1}$ O_2	MS/MS 0.50 mL min $^{-1}$ O_2
Fertilizer 1	12.8 \pm 1.1	8.8 \pm 0.7	8.9 \pm 0.7
Fertilizer 2	36.8 \pm 2.5	30.2 \pm 1.6	29.2 \pm 1.4
Gypsum 1	7.9 \pm 0.04	0.36 \pm 0.01	0.37 \pm 0.01
Gypsum 2	6.4 \pm 1.4	0.71 \pm 0.03	0.65 \pm 0.04

MS/MS: tandem mass spectrometry.

by ICP-MS/MS with proper sensitivity and accuracy. The possibility of a well-controlled conversion of $^{75}\text{As}^+$ ions to $^{75}\text{As}^{16}\text{O}^+$ ions by addition of oxygen gas into the ORS³ cell is one of the main advantages of this instrument configuration, and allows the accurate determination of As in samples with high contents of REEs, such as $^{150}\text{Nd}^{2+}$ and $^{150}\text{Sm}^{2+}$. It can cause false positive errors on As determination causing erroneous decisions about commercialization and application of agricultural products, which is critical considering environmental effects and legislations established for controlling these products.

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