Water adsorption isotherms of coffee blends

Isotermas de adsorção de água em blends de café

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ABSTRACT

This work aimed to evaluate the adsorption isotherms of roasted and grinded coffee, in different blends and storage temperatures. Crude grain coffee of two species were used: arabica (*Coffea arabica*) and robusta(*Coffeacanephora*), dehulled and dried. Grain were submitted to a selection to eliminate deteriorated grain, damaged and bored, to obtain a homogeneous raw material, without defects. Grain coffee from both species, after triage, were sent to roasting process at medium light level (Agtron SCAA#65).After roasting process, grain wasprocessed in a Mahlkönig mill, at average particle size (0.84 m). Blends were made in the following proportions of arabica and robusta: 80:20; 60:40 and 40:60.Samples were stored in BOD chambers for 6 months and kept at temperatures of 10 and 30 °C, with initial moisture content of 2.0 ± 0.3 % (d.b.). At the beginning of storage (time zero) and at 30, 60, 120 and 180 days of storage, moisture content and water activity were determined. Several mathematical models, frequently used to represent the agricultural products hygroscopicity, were fitted to experimental data of sorption. Values of water activity and moisture content varied between 0.1912 and 0.5160; 2.20 and 5.51 % (d.b.), respectively. Modified Oswin model was the one that best represented the adsorption of coffee blends.

Keywords: mathematical modeling, moisture content, water activity

RESUMO

Este trabalho teve como objetivo avaliar as isotermas de adsorção de café torrado e moído, em diferentes formulações de blends e temperaturas de armazenamento. Foram utilizados grãos de café cru de duas espécies: arábica (Coffeaarabica) e robusta (Coffeacanephora), descascados e secos. Os grãos foram submetidos a uma triagem para eliminar os grãos deteriorados, danificados e brocados, a fim de se obter uma matéria-prima homogênea e isenta de defeitos. Os grãos de café de ambas as espécies, após a triagem, foram encaminhados para o processo de torrefação ao nível média clara (Agtron SCAA#65). Após o processo de torrefação, os grãos foram processados em moinho Mahlkönig, na granulometria média (0,84 mm). A formulação dos blends se deu nas seguintes proporções de arábica e robusta: 80:20; 60:40 e 40:60. As amostras foram armazenadas em câmaras BOD durante 6 meses e mantidos às temperaturas de 10 e 30 °C, com teor de água inicial de cerca de 2,0 \pm 0,3 % (b.s.). No início do armazenamento (tempo zero) e aos 30, 60, 120 e 180 dias de armazenamento foram determinadas o teor de água e a atividade de água. Aos dados experimentais de sorção foram ajustados diversos modelos matemáticos frequentemente utilizados para representação da higroscopicidade de produtos agrícolas. Os valores de atividade de água e teor de água variaram entre 0,1912 e 0,5160; 2,20 e 5,51 % (b.s.), respectivamente. O modelo de Oswin modificado foi o que melhor representou a adsorção de blends de café.

Palavras-chave: atividade de água, teor de água, modelagem matemática

1 INTRODUCTION

Brazil is the main producer and exporter of crude grain coffee and the second consumer of coffee in the world, being 6.02 kg of green coffee or 4.82 kg of roasted coffee, per person per year (ABIC, 2018). Two species of greater importance for world commerce of coffee are used: *Coffea arabica* L. and *Coffeacanephora*, known as arabica and robusta coffee, respectively.

In Brazil, arabica coffee is the most produced specie, however, robusta coffee is increasing its share, being used as blends along with arabica coffee (Botelho et al., 2016).Blends among arabica and robusta coffees may be accomplished with the objective to exploit sensorial potential of

both species, combining them to enrich flavor and aroma of the final product, according to the target market.Previous work reported the chemical composition and physical properties of arabica and robusta coffee (Botelho et al., 2016; Corrêaet al., 2016a; Corrêaet al., 2016b; Oliveira et al., 2014; Oliveira et al., 2015), however, it was not found at the specialized literature information regarding the impact of blends over storage.

Storage should be performed properly, in which temperature and relative humidity of the environment surrounding the product are important parameters to be considered during storage. They determine the water activity (a_w) of the product and, thus, the waterexchange between the product, which is hygroscopic, and the environment (Oliveira et al., 2017). This interaction may be studied through the sorption isotherms.

Sorption isotherms are indispensable to determine and analyze water sorption changes during storage and can be expressed by mathematical models, which differ in their theoretical or empirical basis and thenumbers of parameters involved. These models are essential to the prediction and simulation of the performance of materials subjected to a specific process.

The prediction of moisture gain or loss during storage impacts directly at post-harvest procedures for safe storage of the product, affecting its final cost (Corrêaet al., 2016a). Moreover, this information helps to plan actions to prevent and mitigate possible insect and microorganism infestation, as well as assuring higher profitability throughcontrol of prices (Oliveira et al., 2014). Storage of roasted and grinded coffee is not indicated due to grinding promotes a higher interaction between the product and the environment, leading to a higher loss of the constituents, thus, the product quality. However, storage of roasted and grinded coffee may occur due to market difficulties, the need of storage due to lack of transportation, prices that are not suitable its immediate commercialization and the need of blends composition at the industry. Thus, hygroscopic equilibrium data will aid the decision making regarding post-harvest procedures of coffee, roasted and grinded, specially at storage and commercialization stages (Goneliet al., 2013).

Given the above-described considerations and the importance of knowledge of the hygroscopicity of agricultural products, this work had the objective to evaluate the hygroscopic equilibrium of coffee blends, through sorption isotherms, in different storage temperatures, throughout storage for 180 days.

2 MATERIAL AND METHODS

Peeled and dried raw coffee beans of *Coffea arabica* L. and robusta (*Coffeacanephora*) werepurchased at the local market in Manhuaçu, MG. Coffee beans were sorted to remove

deteriorated, damaged, and bored coffee beans to obtain a homogeneous raw material with minimum defects. Afterwards, coffee beans were subjected to the roasting process.

A 300-g raw-coffee capacity preheated, liquefiedpetroleum gas direct roaster with a rotary cylinder operatedat 45 rpm was used to roast the coffee. The degree of roasting was determined by a trained professional by monitoring the samplecolor and comparing it with the Roast Color ClassificationSystem. One roasting degreewasused: medium light (Agtron number SCAA#65). Following the roasting process, the coffee beans wereprocessed in a Mahlkönig mill at medium particle size (0.84 mm).

Blends were then formed at the following proportions of arabica and robusta coffee, respectively: 80:20; 60:40 and 40:60. Samples were then placed in polypropylenebags and refrigerated at biochemical oxygen demand chambersin two storage temperatures (10 and 30 °C). Samples were analyzed at five time points (0, 30,60, 120, and 180 days) over the course of six months, being measured moisture content and water activity.

Moisture content was determined gravimetrically using a forced-air oven at de 105 \pm 1 °C for 24 h, in triplicate (Brasil, 2009). Moisture content of a coffee sample obtained in each storage period (0, 30, 60, 120, and 180 days) wasdefined as the equilibrium moisture content, because the samples were stored in permeable plastic bags and the time elapsed was deemed sufficient to reach equilibrium. Water activity (a_w) of the roasted ground coffee samples were assessed using an AquaLab 4TE water activity meter with an aw accuracy of \pm 0.003. Five replicates were measured.

Mathematical models commonly used to describe sorptionphenomena in agricultural products were fitted to the collected hygroscopic equilibrium data (Table 1).

Model name	Model	Reference		
Copace	$U_e = \exp\left[a - (bT) + (ca_w)\right]$	CORRÊA et al. (1995)	(1)	
Halsey	$U_{e} = \left[\frac{\exp(a-bT)}{-\ln a_{w}}\right]^{1/c}$	HALSEY (1948)	(2)	
Modified Oswin	$U_e = (a + bT) \left(\frac{a_w}{1 - a_w}\right)^{1/c}$	OSWIN (1946)	(3)	
Sigma-Copace	$U_e = \exp\left\{a - (bT) + \left[c \exp\left(a_w\right)\right]\right\}$	CORRÊA et al. (1995)	(4)	

Table 1.Mathematical models used to represent sorptionisotherms

in which: U_e = equilibrium moisture content, % d.b.; a_w = water activity, dimensionless; a, b, c = model parameters, which depends on the product; and, T = temperature, °C.

To fit the mathematical models, a non-linear regression was made, using the Gauss Newton method, with the aid of the *software* STATISTICA $8.0^{\text{®}}$. To assess the suitability of the models, the standard deviation of the estimate (SDE), meanrelative error (MRE) and adjusted determination coefficient (R²) (explained variance) values were analyzed. The adequacy of themodels was also evaluated by the analyses of theresiduals. The SDE and MRE values of each model were calculated using Equations5 and 6.

$$SDE = \sqrt{\frac{\sum_{i=1}^{n} \left(Y - \hat{Y}\right)^{2}}{GLR}}$$
(5)

$$MRE = \frac{100}{n} \sum_{i=1}^{n} \left(\frac{\left| Y - \hat{Y} \right|}{Y} \right)$$
(6)

in which: MRE = mean relative error, %; SDE = standard deviation of the estimate, % d.b. Y = observed value, % d.b.; \hat{Y} = estimated value by the model, % d.b.;n = number of observed data; and, GLR = residue degrees of freedom (number of observeddata minus number of model parameters).

3 RESULTS AND DISCUSSION

Table 2 present the coefficients of the models fitted to the observed data for the hygroscopic equilibrium of coffee beans blends. The corresponding R^2 ,SDE, and MRE values are listed.

Table 2. Parameter estimates of hygroscopic equilibrium models of coffee blends and its respective determination coefficients (R^2), SDE and MRE

Models	Blend	Fitted parameters			MRE	SDE	\mathbb{R}^2	Residual
Models	(arabica:robusta)	a	b	с	(%)	(% d.b.)	(%)	Plot
Copace	80:20	-0.4058 ^{ns}	-0.0067^{ns}	4.1277*	7.91	0.45	88.54	Random
	60:40	0.1808 ^{ns}	-0.0053^{ns}	2.7285^{*}	7.80	0.45	86.47	Random
	40:60	0.3445 ^{ns}	-0.0083 ^{ns}	2.3310^{*}	10.60	0.58	81.43	Random
Halsey	80:20	0.7404^{*}	-0.0044 ^{ns}	0.6646*	7.92	0.45	88.50	Random
	60:40	1.2150^{*}	-0.0054^{ns}	1.0232^{*}	7.84	0.45	86.40	Random
	40:60	1.4549^{*}	-0.0102^{ns}	1.2081^{*}	10.61	0.58	81.40	Random
Modified Oswin	80:20	5.0812^{*}	0.0406^{ns}	1.0610^{*}	7.91	0.45	88.74	Random
	60:40	4.5631*	0.0288^{ns}	1.6524^{*}	7.21	0.43	87.93	Random
	40:60	4.4351*	0.0439 ^{ns}	1.9415^{*}	9.61	0.53	84.02	Random
Sigma-Copace	80:20	-3.0307*	-0.0065 ^{ns}	2.8630^{*}	7.93	0.46	88.29	Random
	60:40	-1.5437*	-0.0050 ^{ns}	1.8847^*	8.59	0.48	84.85	Random
	40:60	-1.1114 ^{ns}	-0.0085 ^{ns}	1.5917^{*}	11.51	0.62	78.77	Random

ns: non-significant by the t-test at 5% of probability

*: significant by the t-test at 5% of probability

At Table 2, it is observed that mathematical modelsused to describe hygroscopicity presented elevated values ofdetermination coefficient (R²), higher than 78%. However, R²is not enough to prove the fitting quality of a non-linearmodel, being used mainly for indicative purposes. Thus, theanalysis of MRE, SDE, and residual plots are required for nonlinearmodels. Regarding the suitability of a specific model fordescribing a phenomenon, MRE values lower than 10% indicatea good fit for practical purposes (Samapundoet al., 2007). The ability of a model to reliably describe a specific physicalprocess is inversely proportional to the SDE value (Draper and Smith, 1998). An analysis of the residuals is commonly performed toensure that a selected model can describe the evaluatedphenomenon, in which random distribution indicate a suitable model to represent the phenomenon (Corrêaet al., 2014).

According to the statistical parameters results, the model that best represents the hygroscopic equilibrium of coffee blends is the Modified Oswin. This model presented MRE values below 10%, lower values of SDE, higher values of R^2 and random distribution of the residuals. Baptestini et al. (2017) also indicated that this model was suitable to predict sorption isotherms of roasted and ground coffee. Figure 1 presents the mean vaues of equilibrium moisture content of coffee blends and its respective sorption isotherms throughout Modified Oswin model, at temperatures of 10 and 30 °C.





В

А



The analysis of the isotherms presented in Figure 1 indicate that the storage temperature affects the hygroscopicity of coffee blends, regardless of the percentage of each coffee species. To reach a given equilibrium moisture content, the water activity should be increased as the temperature increases. Similarly, at a given constant water activity, the equilibrium moisture content decreases with adecrease in storage temperature.

According to Mohsenin(1986), as the temperature increases, molecular vibrations also increase, thereby increasing the distance between molecules and, consequently, decreasing the attractions between molecules, leading to changes of amount of water adsorbed due to temperature alterations at a given relative humidity. Palipaneand Driscoll (1992) have demonstrated that an increase in temperature increases the energy levels of watermolecules, rendering them less thermodynamically stableand, consequently, facilitating the breaking of the bonds between water and sorption sites, which results in a decrease in the moisture content of the product.

A good correspondence was observed between the dataestimated using the Modified Oswinmodel and the observed data (Figure 1). It was not observed a significant difference, during storage, among adsorption isotherms of coffee blends, indicating that the final chemical composition did not altered the hygroscopic equilibrium of coffee blends (Figure 1).

According to the shape-type of the adsorption isotherm presented in Figure 1, these isotherms may be classified as type III isotherms, according to the classification reported by Brunauer(1945). Type III isotherms present a sharp increase of moisture content in its final portion (Figure 1), corresponding to higher values of water activity. This trend was also reported by Oliveira et al. (2017), due to the high capacity of coffee blends to adsorb water from the environment.There is an increase of water activity throughout storage, form 0.192 at the beginning of storage to 0.5196 after 180 days of storage, regardless of coffee blends.

4 CONCLUSIONS

Storage temperature influences directly the hygroscopicity of coffee blends, independently of arabica and robusta proportions. Blends composition did not alter the adsorption isotherms, which presented an increment of moisture content and water activity throughout storage. The Modified Oswin model is the model that represents best the hygroscopicity of coffee blends.

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REFERENCES

Associação Brasileira da Indústria de Café - ABIC. **Indicadores da Indústria de Café**. Disponível em: https://www.abic.com.br/estatisticas/indicadores-da-industria/. Acessoem: 08/04/2020.

BAPTESTINI, F. M.; CORRÊA, P. C.; OLIVEIRA, G. H. H.; CECON, P. R.; SOARES, N. F. F. Kinetic modeling of water sorption by roasted and ground coffee. **Acta Scientiarum. Agronomy,** v.39, n.3, p.273-281, 2017.

BOTELHO, F. M.; CORRÊA, P. C.; BOTELHO, S. C. C.; VARGAS-ELÍAS, G. A.; DINIZ, M. D. M. S.; OLIVEIRA, G. H. H. Propriedades físicas de frutos de café robusta durante a secagem: Determinação e modelagem. **Coffee Science,** v.11, p.65-75, 2016.

BRASIL. **Regras para análise de sementes**.Brasília:Ministério da Agricultura e Reforma Agrária, Secretaria Nacional de defesa Agropecuária, 2009.

BRUNAUER, S. The Adsorption of Gases and Vapors.Princeton: Princeton University Press, 1945.

CORRÊA, P.C.; BOTELHO, F.M.; BOTELHO, S.C.C.; GONELI, A.L.D. Isotermas de sorção de água de frutos de *Coffeacanephora*. **Revista Brasileira de Engenharia Agrícola e Ambiental**, v.18, p.1047-1052, 2014.

CORRÊA, P.C.; MARTINS, D.S.R.; MELO, E.C. **Umigrãos: Programa para o cálculo do teor de umidade de equilíbrio para os principais produtos agrícolas**. Viçosa: Universidade Federal de Viçosa, Centreinar, 1995.

CORRÊA, P. C.; OLIVEIRA, G. H. H.; OLIVEIRA, A. P. L. R.; VARGAS-ELÍAS, G. A.; BAPTESTINI, F. M. Particlesizeandroastingonwatersorption in coniloncoffeeduringstorage. **Coffee Science**, v.11, p.221-233, 2016a.

CORRÊA, P. C.; OLIVEIRA, G. H. H.; OLIVEIRA, A. P. L. R.; VARGAS-ELÍAS, G. A.; SANTOS, F. L.; BAPTESTINI, F. M. Preservation of roasted and ground coffee during storage, part 1: Water content and repose angle. **Revista Brasileira de Engenharia Agrícola e Ambiental**, v.20, p.581-587, 2016b.

DRAPER, N.R.; SMITH, H. Applied regression analysis. New York:Wiley, 1998.

GONELI, A. L. D.; CORRÊA, P. C.; OLIVEIRA, G. H. H.; AFONSO JÚNIOR, P. C. Water sorption properties of coffee fruits, pulped and green coffee. **LWT - Food Science and Technology**, v.50, p.386-391, 2013.

HALSEY, G. Physical adsorption on non-uniform surfaces. Journal of Chemical Physics, v.16, p.931–937, 1948.

MOHSENIN, N. N. **Physical properties of plant and animal materials.**New York:Gordon and Breach Publishers, 1986.

OLIVEIRA, A. P. L. R.; CORRÊA, P. C.; REIS, E. L.; OLIVEIRA, G. H. H. Comparative Study of the Physical and Chemical Characteristics of Coffee and Sensorial Analysis by Principal Components. **Food AnalyticalMethods**, v.8, p.1303-1314, 2015.

OLIVEIRA, G. H. H.; CORRÊA, P. C.; SANTOS, F. L.; VASCONCELOS, W. L.; CALIL JÚNIOR, C.; BAPTESTINI, F. M.; VARGAS-ELÍAS, G. A. Caracterização física de café após torrefação e moagem. **Semina. CiênciasAgrárias**, v.35, p.1813-1828, 2014.

OLIVEIRA, G. H. H.; CORRÊA, P. C.; OLIVEIRA, A. P. L. R.; BAPTESTINI, F. M.; VARGAS-ELIAS, G. A. Roasting, grinding, and storage impact on thermodynamic properties and adsorption isotherms of arabica coffee. **Journal of Food Processing and Preservation**, v.41, p.1-12, 2017.

OSWIN, C. R. The kinetic of package life. III Isotherm. Journal of Chemical Industry, v.65, p.419–421, 1946.

SAMAPUNDO, S.; DEVLIEGHERE, F.; DE MEULENAER, B.; ATUKWASE, A.; LAMBONI, Y.; DEBEVERE, J.M. Sorption isotherms and isosteric heats of sorption of wholeyellow dent corn. **Journal of Food Engineering,** v.79, p.168–175, 2007.