



## Inventory of resources and emissions of cellulose nanostructures from unripe coconut husks: extraction process

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**Abstract** –Inventory of resources and emissions of nanoproducts has been demanded by environmental institutions worldwide to enable life cycle assessment of these products. This work aims to present the preliminary results for the inventory of cellulose nanostructures obtained from coconut husks, considering the extraction process.

Life Cycle Inventory (LCI) of resources and emissions of nanoproducts has been demanded by environmental institutions worldwide to enable life cycle impact assessment of these products. Inventories present measurements of resources (e.g.: kg of materials) and emissions to air, water and soil (e.g.: kg of methane), generated from the raw material extraction to the final disposal stage. The inventoried data is relative to a specific amount of product produced and may be linked to specific environmental impact categories in life cycle impact assessment studies. According to the ISO 14040 [1], inventory studies must define its objective, functional unit and elementary processes considered.

This inventory is part of the life cycle assessment (LCA) study of cellulose nanostructures from unripe coconut fiber that includes three life cycle stages: unripe coconut husks collection and transport, fiber extraction, and nanostructure extraction. The data presented in this work is related only to resources and emissions occurring in the stage of nanostructure extraction (gate to gate LCI). The function of cellulose nanostructures is defined as improvement of mechanical properties of composites and the function unit adopted is the production of 1g of cellulose nanostructures from unripe coconut husks fiber.

The elementary processes involved in the extraction of cellulose nanostructures are milling of coconut husk fibers, washing to remove impurities and waxy substances, bleaching to start removing lignin, acid hydrolysis to obtain nanostructures, centrifugation to separate nanostructures from the acid solution and dialysis to correct the pH of the solution containing nanostructures. The detailed description of this process is presented by Rosa et al [2]. The measurements of resources and emissions were carried out when this process was performed five times in the laboratories of Embrapa Tropical Industry.

The preliminary results of the inventory are presented in Table 1. The studied process presents an efficiency of 6% related to the conversion of unripe coconut fiber into cellulose nanostructures. Considering that the cellulose content of unripe coconut fiber is around 38% [2], there is opportunities for improvements in process efficiency.

The next step of this work is the identification of these opportunities, through the analysis of data in each elementary process. A comparative study will also be performed considering resources and consumptions related to the production of cellulose nanostructures obtained from cotton and sugar-cane fibers. This work has been supported by Embrapa's Nanotechnology Network (Agronano).

**Table 1:** Resources and emissions for the production of 1g of cellulose nanostructures.

| Inputs               | Unit | Average value | Standard deviation | Outputs                           | Unit | Average value | Standard deviation |
|----------------------|------|---------------|--------------------|-----------------------------------|------|---------------|--------------------|
| Unripe coconut fiber | g    | 15.89         | 0.67               | Emissions to soil                 |      |               |                    |
| Water                | L    | 21.60         | 2.04               | Residues of unripe coconut fibers | g    | 6,66          | 0,44               |
| Electricity          | kWh  | 3.95          | 0.48               | Emissions to water                |      |               |                    |
| Sodium hydroxide     | g    | 1.72          | 0.01               | Electric Conductivity (EC)        | dS/m | 39,909        | 77,742             |
| Acetic acid          | g    | 0.78          | 0.05               | Oxygen Chemical Demand (DQO)      | g    | 5,519         | 0,648              |
| Sulfuric acid        | g    | 26.33         | 2.60               | Oxygen Biochemical Demand (DBO)   | g    | 2,693         | 0,310              |
| Sodium chlorite      | g    | 1.54          | 0.10               | Nitrate                           | g    | 0,017         | 0,002              |
| Nitric acid          | g    | 4.25          | 0.30               | Total Kjeldahl Nitrogen (NT)      | g    | 0,008         | 0,001              |
|                      |      |               |                    | Phenol                            | g    | 0,047         | 0,004              |
|                      |      |               |                    | Total Phosphorous                 | g    | 0,0037        | 0,0003             |
|                      |      |               |                    | pH                                |      | 4,731         | 4,276              |
|                      |      |               |                    | Total Suspended Solids (TSS)      | g    | 0,733         | 0,088              |
|                      |      |               |                    | Electric Conductivity (EC)        | dS/m | 39,909        | 77,742             |
|                      |      |               |                    | Oxygen Chemical Demand (DQO)      | g    | 5,519         | 0,648              |

### References

- [1] Associação Brasileira de Normas Técnicas (ABNT). NBR ISO 14040, Rio de Janeiro, ABNT, 2009.  
[2] M.F. Rosa, E.S. Medeiros, J.A. Malmonge, K.S. Gregorski, D.F. Wood, L.H.C. Mattoso, G. Glenn, W.J. Orts b, S.H. Imamb, Cellulose nanowhiskers from coconut husk fibers: Effect of preparation conditions on their thermal and morphological behavior. Carbohydrate Polymers, 2010.