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Société Française
de Métallurgie et de Matériaux



Associazione Italiana
di Metallurgia

Main Topics

- Functional Materials: Properties & Applications**
Magnetic Materials; Materials for Nanostructures; Functional Polymeric Materials;
MEMS, NEMS and Devices
- Structural Materials: Properties & Applications**
Advanced Metals; Advanced Ceramics; Hybrid and Composite Materials;
Advanced Concepts in Structural Materials
- Materials Processing**
Solidification and Solid State Transformations; Joining; Powder Routes: from Synthesis
to Materials; Coatings and Surface Engineering; New concepts in Materials Processing
- Characterisation and Modeling**
Microstructural Characterisation Techniques; Mechanical Characterisation
Techniques ; Materials Modeling on all Length Scales
- Energy and related Applications**
Energy Production, Transportation and Management; Materials for Energy
in a Sustainable Society; Materials for Transportation
- Health Care Applications**
Materials for Health Care Applications
- Education

Topic Area	C Processing	
Topic	C3 - Powder routes : from synthesis to materials	C31-O-4-5
Symposium	C31 - Powder Synthesis and Processing	2444
Session	Ceramic powders synthesis II	

Synthesis of morphologically different TiO₂-anatase and the study of its photocatalytic properties.

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Abstract

In studies on the influence of the nanocrystals morphology on its photocatalytic properties, it is important to consider the historical of the material, like the synthesis process.[1] This paper proposes to investigate the influence of the synthesis process on the photoactivity of morphologically different TiO₂ nanocrystals obtained by hydrothermal treatment of amorphous powder at different pH conditions.[2,3] Photoactivity of the samples was studied by the degradation of Rhodamine b (RhB) dye under UVC radiation. Nanocrystals were synthesized at pH 6, 8 and 10, which resulted in anisotropic materials, because of the oriented crystal growth with higher surface hydroxylation to the samples synthesized at higher pH values, which is favorable to photocatalysis. Photodegradation trials showed the strong influence of morphology on the photoactivity of the material. The trials also showed that synthesis performed at pH value around 8 generated materials with higher photoactivities than at other pH values. The transformation of isotropic structures (pH 6 and 8) to anisotropic (pH 10) leads to a decrease in the photoactivity of the material, even the last one having a higher surface hydroxylation. One of the factors that lead to decrease in photoactivity is the decrease in specific surface area, however this is not the only factor. Photoluminescence spectroscopy showed relation with synthesis pH. Anisotropic crystal growth generated a decrease in spectra intensity. It was because of defects that coming from crystal growth. These defects act as recombination centers of photogenerated charges during photocatalytic process, leading to a non-radiative decay and preventing electrons and holes responsible for the photocatalytic process to reach the surface, resulting in a decrease in photoactivity materials synthesized at higher pH values, even those possessing a higher surface hydroxylation.

- 1- W.Y. Teoh, F. Denny, R. Amal, D. Friedmann, L. Madler, S.E. Pratsinis, *Top. Catal.* 44 (2007) 489.
- 2- C. Ribeiro, C.M. Barrado, E.R. Camargo, E. Longo, E.R. Leite, *Chemistry - A European Journal* 15 (2009) 2217.
- 3- S. Han, S.H. Choi, S.S. Kim, M. Cho, B. Jang, D.Y. Kim, J. Yoon, T. Hyeon, *Small* 1 (2005) 812.

